

Convenient one-step synthesis of a medicinally relevant benzopyranopyridine system

Nikolai M. Evdokimov,^a Artem S. Kireev,^b Andrey A. Yakovenko,^c
Mikhail Yu. Antipin,^{c,d} Igor V. Magedov^{a,e,*} and Alexander Kornienko^{b,*}

^aDepartment of Organic Chemistry, Timiryazev Agriculture Academy, Moscow 127550, Russia

^bDepartment of Chemistry, New Mexico Institute of Mining and Technology, Socorro, New Mexico 87801, USA

^cDepartment of Natural Sciences, New Mexico Highlands University, Las Vegas, New Mexico 87701, USA

^dInstitute of Organoelement Compounds, Russian Academy of Sciences, Moscow, Russia

^eIntelbioscan Ltd., Timiryazevsky Proesd 2, Moscow 127550, Russia

Received 9 October 2006; revised 18 October 2006; accepted 19 October 2006

Available online 9 November 2006

Abstract—Benzopyrano[2,3-*b*]pyridine is an important privileged medicinal scaffold. A three-component reaction of salicylaldehydes, thiols and 2 equiv of malononitrile that leads to the formation of a series of compounds incorporating 2,4-diamino-3-cyano-5-sulfanylbenzopyrano[2,3-*b*]pyridine framework is described. A proposed mechanism with the supporting experimental data is presented.

© 2006 Elsevier Ltd. All rights reserved.

The rapid assembly of molecular diversity utilizing multi-component reactions has received a great deal of attention, most notably for the construction of heterocyclic ‘drug-like’ libraries.¹ These methodologies are of particularly great utility when they lead to the formation of ‘privileged medicinal scaffolds,’ defined as molecular frameworks serving as the basis for the generation of ligands for functionally and structurally discreet biological receptors.² Such chemistry greatly facilitates the development of pharmaceutical agents for diverse applications.

Benzopyrano[2,3-*b*]pyridine scaffold is of a significant medicinal relevance. The examples of approved therapeutic agents incorporating this molecular framework include amlexanox and pranoprofen (Fig. 1).

In addition, many of these compounds possess anti-proliferative,³ cancer chemopreventive,⁴ anti-bacterial (including anti-tubercular),⁵ anti-myopic,⁶ anti-histaminic,⁷ hypotensive,⁸ anti-rheumatic⁹ and anti-asthmatic

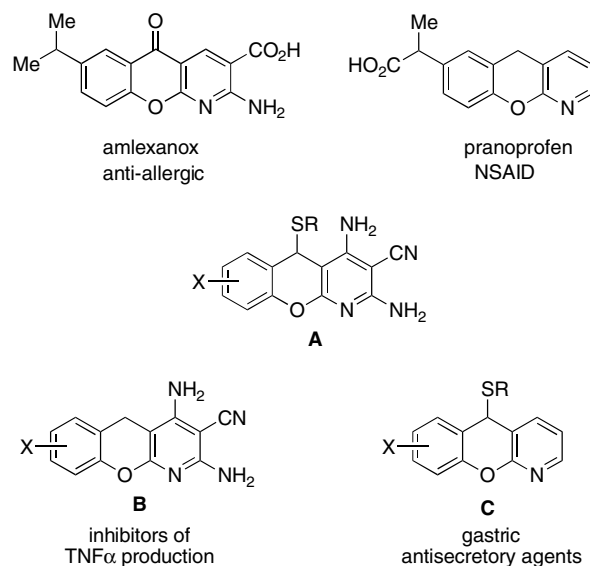


Figure 1.

activities.¹⁰ Many synthetic pathways to such medicinal libraries have been reported.¹¹ However, the diverse pharmacological properties associated with benzopyranopyridines warrant the development of novel processes allowing the synthesis of previously inaccessible

Keywords: Multicomponent reaction; Privileged medicinal scaffold; Drug-like heterocycle.

* Corresponding authors. Tel.: +1 505 835 5884; fax: +1 505 835 5364 (A.K.); e-mail addresses: intelbioscan@mtu-net.ru; akornien@nmt.edu

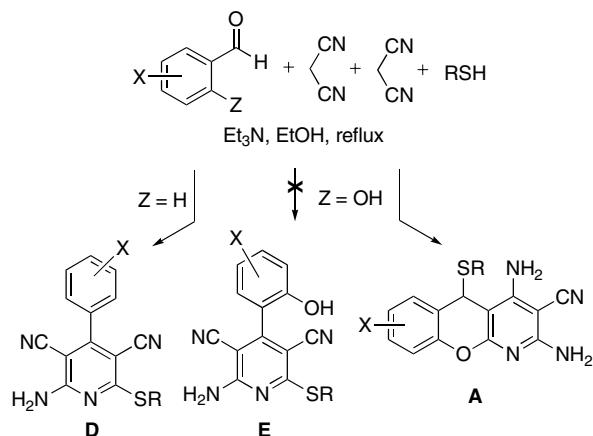


Figure 2.

analogues for biological evaluation. In this letter, we describe a multi-component strategy for the rapid preparation of library **A** (Fig. 1). Examples of structurally relevant biologically active compounds are benzopyranopyridines **B**, found to inhibit mitogen-activated protein kinase-activated protein kinase 2 and attenuate the production of pro-inflammatory $\text{TNF}\alpha$,¹² and **C**, reported to inhibit histamine-stimulated gastric acid secretion in animals.¹³

We previously disclosed a one-step three-component synthesis of 3,5-dicyanopyridines **D** starting from various aldehydes, thiols and malononitrile (Fig. 2).¹⁴

In an attempt to extend the method for the preparation of pyridines **E** we employed salicylic aldehydes and obtained a library of compounds whose elemental analysis data and the NMR spectra were inconsistent with the expected structures. The X-ray analysis showed these compounds to have a benzopyrano[2,3-*b*]pyridine framework **A** (Fig. 3).¹⁵

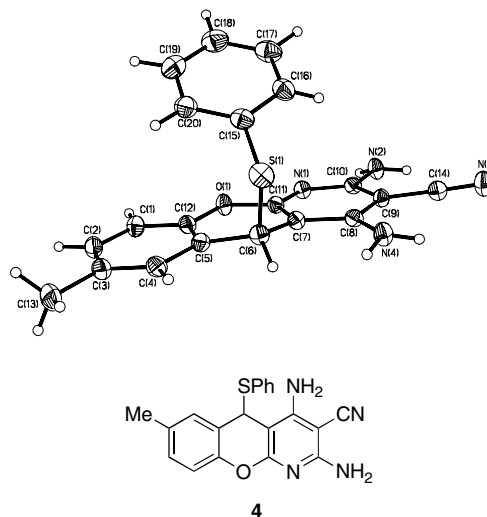


Figure 3.

The reaction works well for all salicylaldehyde and thiol combinations tested. The products precipitate from refluxing ethanolic solutions and are isolated by simple filtration. The yields of recrystallized benzopyranopyridines are given in Table 1.^{16,17}

Our proposed mechanistic interpretation of the divergence in reaction paths for *o,o*-unsubstituted aldehydes, leading to formation of pyridines **D**, and salicylic aldehydes, resulting in benzopyranopyridines **A**, is shown in Figure 4.

Base-catalyzed Michael addition of thiols to Knoevenagel adducts **F**, produced from *o,o*-unsubstituted aldehydes, to form **G** is reversible and non-productive. However, the formal assembly process **I** leads irreversibly to the formation of dihydropyridines **K**, thermodynamically stabilized by the push–pull interactions of the

Table 1. One-step synthesis of benzopyranopyridines

Benzopyranopyridine	R ₁	R ₂	R ₃	R	Yield (%)
1	H	H	H	Ph	86
2	H	H	H		55
3	H	H	H	PhCH ₂	51
4	H	H	Me	Ph	69
5	H	H	Me	PhCH ₂	52
6	H	H	Me		63
7	H	H	Me	Mes	56
8	OMe	H	H	Ph	57
9	H	Et ₂ N	H	Ph	63
10	H	OH	H	Ph	63
11	H	H	NO ₂	4-F-Ph	73

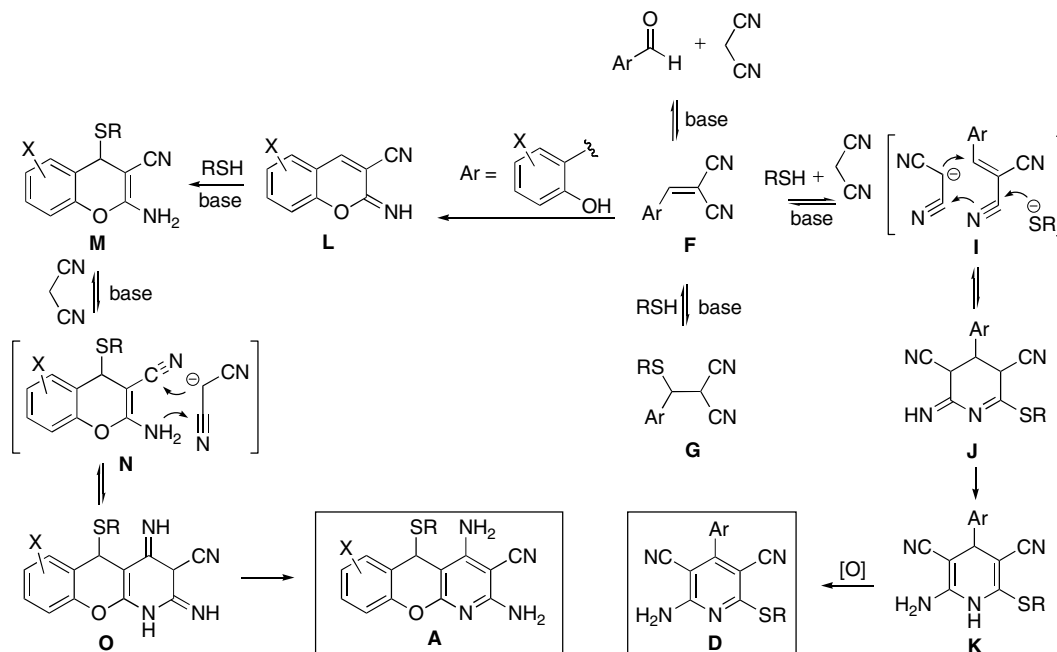


Figure 4.

donor (positions 2 and 6) and acceptor (positions 3 and 5) substituents. Oxidative aromatization then affords pyridines **D**. In contrast, Knoevenagel intermediates **F**, produced from salicylaldehydes, undergo intramolecular cyclization to give powerful Michael acceptors **L** that react with strongly nucleophilic thiols to form thermodynamically stable chromenes **M** irreversibly. Lastly, the addition of another equivalent of malononitrile results in benzopyranopyridines **A**.

In support of the proposed mechanism we obtained experimental evidence for each of the key steps in Figure 4. Thus, *o*-hydroxybenzaldehyde reacts with 1 equiv of malononitrile to form iminochromene **12** in a 64% isolated yield after recrystallization (Fig. 5). This compound undergoes an addition with thiophenol to afford phenylsulfanylchromene **13** in a quantitative yield. Finally, when **13** is treated under the same reaction conditions with another equivalent of malononitrile, benzopyranopyridine **1** forms in a good isolated yield.

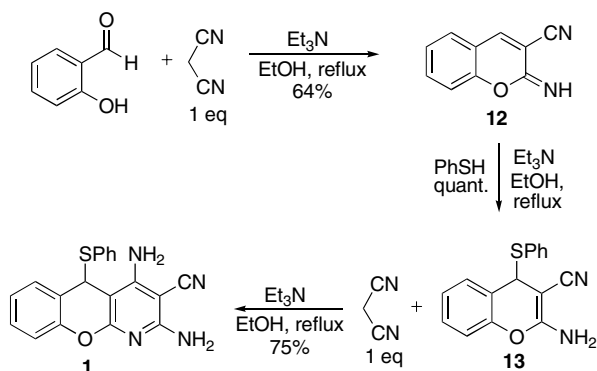


Figure 5.

Further exploration of this chemistry and biological testing of the synthesized benzopyranopyridines are in progress and will be reported in due course.

Acknowledgements

A.K. thanks the US National Institutes of Health (CA-99957 and RR-16480) for financial support of this work. A.A.Y. and M.Yu.A. are grateful to NSF/DMR (Grant 0420863) for the acquisition of X-ray single crystal diffractometer and to the Distributed Nanomaterials Characterization Network in the framework of New Mexico NSF EPSCoR Nanoscience initiative.

References and notes

- For recent reviews, see: (a) Gerencsér, J.; Dormán, G.; Darvas, F. *QSAR Comb. Sci.* **2006**, 439–448; (b) Ramón, D. J.; Yus, M. *Angew. Chem., Int. Ed.* **2005**, 44, 1602–1634; (c) Orru, R. V. A.; de Greef, M. *Synthesis* **2003**, 1471–1499; (d) Hulme, C.; Gore, V. *Curr. Med. Chem.* **2003**, 10, 51–80; (e) Ugi, I.; Heck, S. *Comb. Chem. High Throughput Screen* **2001**, 4, 1–34; (f) Weber, L.; Illgen, K.; Almstetter, M. *Synlett* **1999**, 366–374.
- The term ‘privileged scaffolds or structures’ was originally introduced by Merck researchers in their work on benzodiazepines: (a) Evans, B. E.; Rittle, K. E.; Bock, M. G.; DiPardo, R. M.; Freidinger, R. M.; Whitter, W. L.; Lundell, G. F.; Veber, D. F.; Anderson, P. S.; Chang, R. S. L.; Lotti, V. J.; Cerino, D. J.; Chen, T. B.; Kling, P. J.; Kunkel, K. A.; Springer, J. P.; Hirshfield, J. *J. Med. Chem.* **1988**, 31, 2235–2246; For a review, see: (b) Patchett, A. A.; Nargund, R. P. *Ann. Rep. Med. Chem.* **2000**, 35, 289–298.
- Kolokythas, G.; Pouli, N.; Marakos, P.; Pratsinis, H.; Kletsas, D. *Eur. J. Med. Chem.* **2006**, 41, 71–79.

- Azuine, M. A.; Tokuda, H.; Takayasu, J.; Enjyo, F.; Mukainaka, T.; Konoshima, T.; Nishino, H.; Kapadia, G. *J. Pharmacol. Res.* **2004**, *49*, 161–169.
- (a) Srivastava, S. K.; Tripathi, R. P.; Ramachandran, R. *J. Biol. Chem.* **2005**, *280*, 30273–30281; (b) Brötz-Oesterhelt, H.; Knezevic, I.; Bartel, S.; Lampe, T.; Warnecke-Eberz, U.; Ziegelbauer, K.; Häbich, D.; Labischinski, H. *J. Biol. Chem.* **2003**, *278*, 39435–39442.
- Toshiro, S.; Noriko, W. Eur. Pat. Appl. EP 647445 A1 19950412, 1995.
- Ito, Y.; Kato, H.; Yasuda, S.; Kato, N.; Iwasaki, N.; Nishino, H.; Takeshita, M. Jpn. Kokai Tokkyo Koho, JP 06107664 A2 19940419, 1994.
- Goto, K.; Yaoka, O.; Oe, T. PCT Int. Appl. WO 8401711 A1 19840510, 1984.
- Maruyama, Y.; Goto, K.; Terasawa, M. Ger. Offen. DE 3010751 19810806, 1981.
- Ukawa, K.; Ishiguro, T.; Kuriki, H.; Nohara, A. *Chem. Pharm. Bull.* **1985**, *33*, 4432–4437.
- For recent synthetic work, see: (a) Abdel-Rahman, A. H.; Hammouda, M. A. A.; El-Desoky, S. I. *Heteroat. Chem.* **2005**, *16*, 20–27; (b) Langer, P.; Appel, B. *Tetrahedron Lett.* **2003**, *44*, 5133–5135; (c) Daia, D. E.; Gabbutt, C. D.; Heron, B. M.; Hepworth, J. D.; Hursthouse, M. B.; Abdul Malik, K. M. *Tetrahedron Lett.* **2003**, *44*, 1461–1464; (d) Fujiwara, H.; Kitagawa, K. *Heterocycles* **2000**, *53*, 409–418; (e) O'Callaghan, C. N.; McMurry, T. B. H.; O'Brien, J. E.; Draper, S. M. *J. Chem. Res. (S)* **1997**, 312–313; (f) O'Callaghan, C. N.; McMurry, T. B. H.; O'Brien *J. Chem. Soc., Perkin Trans. 1* **1995**, 417–420.
- Anderson, D. R.; Hegde, S.; Reinhard, E.; Gomez, L.; Vernier, W. F.; Lee, L.; Liu, S.; Sambandam, A.; Snider, P. A.; Masih, L. *Bioorg. Med. Chem. Lett.* **2005**, *15*, 1587–1590.
- Bristol, J. A.; Gold, E. H.; Gross, I.; Lovey, R. G.; Long, J. F. *J. Med. Chem.* **1981**, *24*, 1010–1013.
- Evdokimov, N. M.; Magedov, I. V.; Kireev, A. S.; Kornienko, A. *Org. Lett.* **2006**, *8*, 899–902.
- Crystallographic data for compound **4**: $C_{20}H_{16}N_4OS$, $M_r = 360.43$, triclinic space group $P\bar{1}$, $a = 7.3717(6)$, $b = 9.5840(7)$, $c = 12.8924(10)$ Å, $\alpha = 103.820(2)$, $\beta = 92.975(2)$, $\gamma = 99.441(2)^\circ$, $V = 868.55(12)$ Å³, $Z = 2$, $T = 120(2)$ K, $F(000) = 376$, $D_{\text{calcd}} = 1.378$ g cm⁻³, $\theta_{\text{max}} = 29.99^\circ$, 9671 reflections measured and 4882 unique ($R_{\text{int}} = 0.0190$) reflections, full matrix least-squares refinement on F^2 , R_1 (obs) = 0.0494, and wR_2 (all data) = 0.1037. Supplementary data in the form of CIFs have been deposited with the Cambridge Crystallographic Data Centre (CCDC 622796). Copies of the data can be obtained, free of charge, on application to CCDC, 12 Union Road, Cambridge CB2 1EZ, UK [fax: +44 1223 336033 or e-mail: deposit@ccdc.cam.ac.uk].
- General procedure for benzopyranopyridine synthesis*: To a mixture of a selected salicylaldehyde (1.5 mmol), malononitrile (3 mmol) and a desired thiol (1.5 mmol) in 7 mL of anhydrous ethanol was added Et₃N (0.1 mmol) dropwise at room temperature. The resulting mixture was refluxed for 3–3.5 h and then allowed to cool to room temperature. The formed precipitate was isolated by filtration. The product was dissolved in DMF (3 mL), and the remaining undissolved material was removed by filtration. To the filtrate was added water (4 mL), which resulted in the crystallization of the product. The formed crystals were isolated by filtration to yield a corresponding pure benzopyranopyridine (**1–11**).
- Selected characterization data*: **4**: 69%; mp 275–276 °C (DMF); ¹H NMR (DMSO-*d*₆) δ 2.24 (s, 3H), 5.69 (s, 1H), 6.32 (br s, 2H), 6.72 (d, $J = 8.2$ Hz, 1H), 6.78 (br s, 2H), 6.83 (dd, $J = 1.3, 7.3$ Hz, 2H), 6.91 (d, $J = 1.5$ Hz, 1H), 7.01 (dd, $J = 1.5, 8.2$ Hz, 1H), 7.13 (t, $J = 7.3$ Hz, 2H), 7.28 (dd, $J = 1.2, 7.3$ Hz, 1H); ¹³C NMR (DMSO-*d*₆) δ 160.2, 160.0, 156.9, 149.2, 136.6, 132.9, 131.2, 129.5, 129.4, 129.2, 129.0, 128.6, 121.8, 116.9, 116.0, 86.6, 70.6, 43.3, 20.7; IR (KBr) 3480, 3410, 3374, 3144, 2926, 2204, 1658, 1616, 1500, 1478, 1398, 1332, 1262, 1226, 1164, 1076, 1024, 900, 820, 780, 740, 692, 644, 532 cm⁻¹. Anal. Calcd for $C_{20}H_{16}N_4OS$ (360.442): C, 66.64; H, 4.48; N, 15.55; S, 8.89. Found: C, 66.51; H, 4.52; N, 15.68; S, 8.83. Compound **6**: 63%; mp 240–242 °C (DMF); ¹H NMR (DMSO-*d*₆) δ 2.31 (s, 3H), 3.50 (d, $J = 14.0$ Hz, 1H), 3.57 (d, $J = 14.0$ Hz, 1H), 5.43 (s, 1H), 6.02 (dd, $J = 0.9, 3.0$ Hz, 1H), 6.26 (d, $J = 1.8$ Hz, 1H), 6.39 (br s, 2H), 6.65 (br s, 2H), 6.98 (d, $J = 8.8$ Hz, 1H), 7.09 (d, $J = 1.8$ Hz, 1H), 7.11 (dd, $J = 1.8, 8.8$ Hz, 1H), 7.41 (dd, $J = 0.9, 1.8$ Hz, 1H); ¹³C NMR (DMSO-*d*₆) δ 160.7, 160.3, 157.1, 151.2, 149.4, 142.7, 133.8, 129.7, 129.0, 116.9, 116.5, 111.0, 107.8, 87.1, 71.0, 25.9, 20.9; IR (KBr) 3458, 3350, 3240, 3160, 2918, 2200, 1628, 1606, 1575, 1498, 1472, 1398, 1332, 1262, 1226, 1155, 1008, 780, 745, 538 cm⁻¹. Anal. Calcd for $C_{19}H_{16}N_4O_2S$ (364.431): C, 62.62; H, 4.43; N, 15.38; S, 8.80. Found: C, 62.45; H, 4.31; N, 15.52; S, 8.91. Compound **8**: 57%; mp 254–255 °C (DMF); ¹H NMR (DMSO-*d*₆) δ 3.72 (s, 3H), 5.74 (s, 1H), 6.30 (br s, 2H), 6.74 (dd, $J = 1.4, 7.9$ Hz, 1H), 6.80 (br s, 2H), 6.84 (dd, $J = 1.2, 7.3$ Hz, 2H), 6.90 (dd, $J = 1.4, 7.9$ Hz, 1H), 7.01 (t, $J = 7.9$ Hz, 1H), 7.12 (t, $J = 7.3$ Hz, 2H), 7.28 (dt, $J = 1.2, 7.3$ Hz, 1H); ¹³C NMR (DMSO-*d*₆) δ 160.3, 160.1, 156.9, 147.5, 141.3, 136.4, 131.3, 129.4, 128.8, 123.9, 123.0, 120.5, 117.1, 111.9, 86.7, 70.9, 56.5, 43.4; IR (KBr) 3452, 3340, 3272, 2866, 2208, 1656, 1630, 1606, 1590, 1486, 1408, 1350, 1270, 1220, 1172, 1116, 1096, 794, 778, 764, 748, 710 cm⁻¹. Anal. Calcd for $C_{20}H_{16}N_4O_2S$ (376.442): C, 63.81; H, 4.29; N, 14.89; S, 8.52. Found: C, 63.95; H 4.08; N, 14.81; S, 8.63.